

Appl. No.: 10/821,426  
Amdt. dated: October 5, 2006  
Reply to the Office action mailed: July 31, 2006

**Amendments to the Specification:**

Please replace the third full paragraph on page 5 with the following amended paragraph:

The concentration of probucol or its derivatives in the above salt-forming reaction is dramatically higher than that disclosed in similar prior art reactions. Thus, in carrying out the salt-forming reaction, the ratio to solvent to probucol derivative by weight is from about 2:1 to about 1:5, preferably about 1:1 to about 3:10, most preferably 3:5.

Please replace the last paragraph on page 8 bridging to page 9 of the specification with the following new paragraph:

~~The solution of the compounds of~~ The reaction mixture of the compounds of Formula 2 is acidified in the presence of the same or a different organic hydrocarbon solvent that was previously used to remove the unreacted probucol or probucol derivative, ~~i.e., a hydrocarbon solvent having the formula  $C_nH_{2n+2}$  where n is an integer from 5 to 12.~~ In this way, the pH of such reaction mixture formed in reaction step (2) is reduced to less than 7 and then an organic hydrocarbon solvent having the formula  $C_nH_{2n+2}$  where n is an integer from 5 to 12 is added to the reduced pH reaction mixture. The hydrocarbon solvent preferentially dissolves the compounds of Formula 2 where Z and Z' are different and are hydrogen and the moiety - C(O) - C<sub>1</sub> to C<sub>6</sub> alkyl C(O)OH or the moiety C(O) - C<sub>1</sub> to C<sub>6</sub> alkenyl C(O)OH where alkyl and alkenyl are as previously defined.